

DATA GAPS RELATED TO THERMODYNAMICS IDENTIFIED IN EURAD-FUTURE PROJECT

EURAD2 – annual event 1, Bologna

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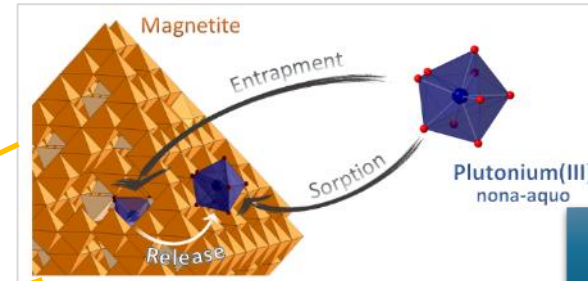
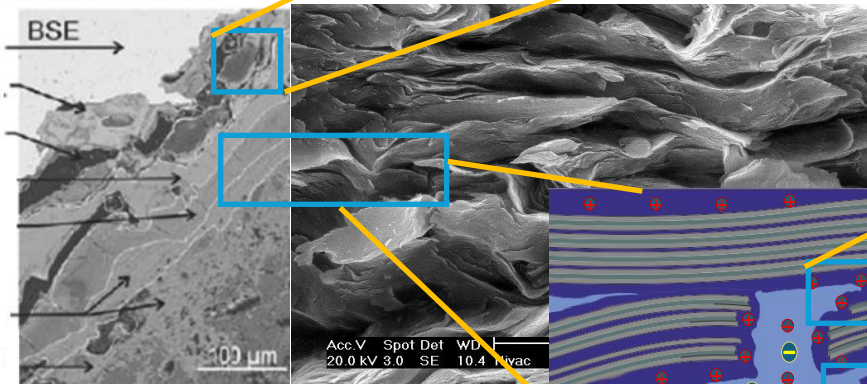
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FUNDAMENTAL UNDERSTANDING OF RADIONUCLIDE RETENTION

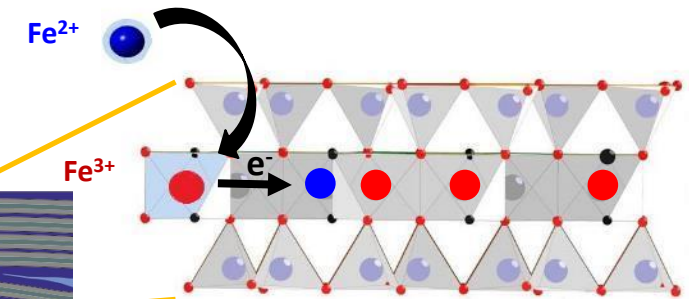
Pillars of RN transport and retention

Argillaceous Rocks

Fe metal
Magnetite
Chukanovite
Fe^{II} phyllosilicate
Ca-doped siderite
Clay

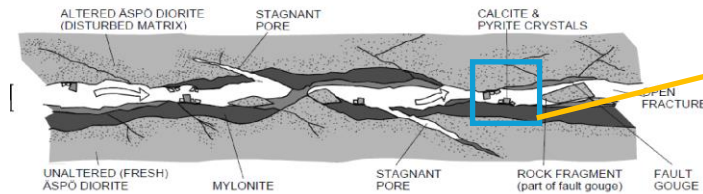


WP 3.0 REDOX



WP 2.3 SORPTION
REVERSIBILITY

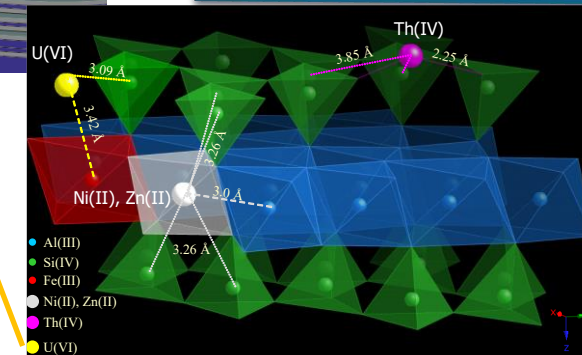
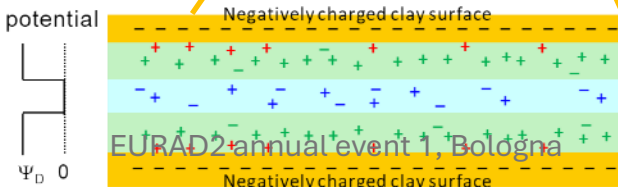
Crystalline Rocks



WP MOBILITY:
2.1 Clay & 2.2 Crystalline

Profile of electrical
potential

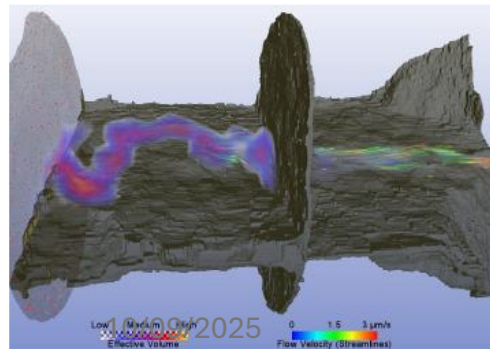
Clay
Pore



EDL → J_{i,DL}

Free water → J_{i,free}

EDL → J_{i,DL}



eurad2



AIM OF FUTURE:

DELIVER SOLID SCIENTIFIC KNOWLEDGE ON RN MOBILITY TO SUPPORT SAFETY ASSESSMENT

- ☐ **Obtain mechanistic understanding of radionuclide mobility and retention w.r.t.**
 - ☐ **EDL effects:** anion exclusion, enhanced flux in EDL for cations
 - ☐ **microstructural effects** (water saturation, pore heterogeneity and connectivity, grain boundaries)
 - ☐ **Chemical aspects** of pore water composition (pH, IS, competing elements)
 - ☐ **Interplay of transport regime and chemical boundary conditions** in fractured crystalline rocks

- ☐ **Understand the nature of the long-term retention mechanisms i.e. adsorption reversibility**
 - ☐ Adsorption versus incorporation (surface precipitation, neo-formation), competitive adsorption, solid solutions

- ☐ **Quantify/Characterize redox controlled retention and mobility processes**
 - ☐ Interaction between redox-sensitive radionuclides and relevant Fe-bearing minerals (clays & oxides)
 - ☐ Structural constraints on availability of redox active elements in the system: e.g. total Fe content, structural location and molecular scale surface speciation

HOW IS THIS ACHIEVED?

❑ Systems Studied:

- ❑ Model systems: Pure clay minerals (Ill, MoMo,...), fracture filling materials (calcite, chlorite,...) & Fe-bearing clays&minerals (Notronite, magnetite, green rust,...)
- ❑ Natural systems: argillaceous (COx; OPA, Boda) and crystalline rocks (Bukov, Onkalo)
- ❑ Weakly, moderately and strongly sorbing + redox sensitive radionuclides (I, Se, Ra, Ba, Co, Ni, Zn, Tc, Re, U, Np, Pu, Am,...)

❑ Experiments

- ❑ Sorption experiments (disperse/compacted) under highly controlled conditions
- ❑ Transport experiments (different scales) under highly controlled conditions
- ❑ (Cutting edge) Spectroscopy (Mossbauer, XPS, synchrotron-XAS techniques, ATR-FTIR, LA-ICP-MS, TOF-SIMS/rL-SNMS, AMS, 14C-autoradiography, PET)

THERMODYNAMIC DATA IN FUTURE – ROLE AND IMPACT?

- ❑ ***No explicit focus on gathering or validating thermodynamic data - focus on process understanding***

- ❑ Experimental conditions (often) tailored towards this goal:
choice of RN, material, availability of (thermodynamic) data...

- ❑ ***Implicit use of thermodynamic data***

- ❑ Design of experiments e.g. speciation calculations, predictive calculations, controlling conditions,...
 - ❑ Modelling of sorption(interaction) experiments
 - ❑ Reactive transport modelling of transport experiments
 - ❑ Interpretation of spectroscopic data

→ **high quality thermodynamic data were key to the success of the FUTURE project**

BIRD EYE CONCLUSIONS AND OUTCOME OF THE PROJECT

- ❑ **Validity of *bottom up* approach for sorption is broadly confirmed but not always straightforward ... as all underlying mechanisms and parameters must be known:**
 - ❑ From dispersed systems to compacted systems -> **straightforward if system chemistry is well known**
 - ❑ From single mineral to complex systems -> **straightforward**
 - ❑ From single RN to RN mix (competition) -> **remains challenging** and **needs further studies**
- ❑ **New/missing sorption data became available and included into sorption models**
- ❑ **Sorption reversibility should not be taken for granted... often some irreversibility is observed but mechanisms are difficult to unravel**
- ❑ **Mechanistic sorption models are applicable in the interpretation of transport experiments (but PW chemistry should be accounted for)**
- ❑ **Mechanistic understanding of anion exclusion and surface diffusion has matured**
- ❑ **Development of spectroscopic techniques helped to:**
 - ❑ Push detection limits for concentration measurements by several orders of magnitude
 - ❑ Provide detailed zoom into surface properties
 - ❑ Revealed quantitative information about the RN speciation in aqueous phase, in solids and at interfaces

GAINED UNDERSTANDING (DATA AND MECHANISMS) HELPS TO IMPROVE MODEL DESCRIPTIONS DESCRIBING RADIONUCLIDE TRANSPORT AT DIFFERENT SCALES SUITABLE FOR SA PURPOSES

- ☐ Extending datasets for sorption and diffusion on rocks and compacted minerals
 - Support for the selection and scientific justification of (K_d , D_e ...) values in the SA
- ☐ Clear evidence for enhanced cation diffusion in clay systems (surface diffusion)
 - Must be considered in generalized SA
- ☐ New data for anions transport and anion exclusion: divalent anions, particle orientation, partial saturation, ...
 - current SA model can be refined
- ☐ Significant improvement in understanding the mechanisms of redox processes in which iron bearing phases and clays can promote sorption-reduction reactions
 - Redox controlled uptake in the vicinity of wastes by corrosion products should be considered in SA
 - Effect of redox-mediated processes on radionuclide mobility needs further investigations.
- ☐ Limitations of current retention/diffusion models (leading to overestimation and underestimation) are clearly identified:
 - Competition, slow kinetic process, sorption onto non-clays minerals, solid solution formations



KNOWLEDGE GAPS RELATED TO THERMODYNAMICS

☐ **Kinetic data**

- ☐ Redox processes
- ☐ Retention mechanisms other than reversible sorption

☐ **Completeness of thermodynamic databases**

- ☐ Scarcity of data for some elements (experimental difficulties of all kinds)
- ☐ Incomplete data for some elements
- ☐ Internal consistency

☐ **Thermodynamics in confined systems**

☐ **Solid solution thermodynamic aspects**