



## Radionuclide migration studies performed with the Task 4 of CORI workpackage

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NUWCEM 2022 - 4th International Symposium on Cement-based Materials for Nuclear wastes, May 4 to 6, 2022, Avignon, France

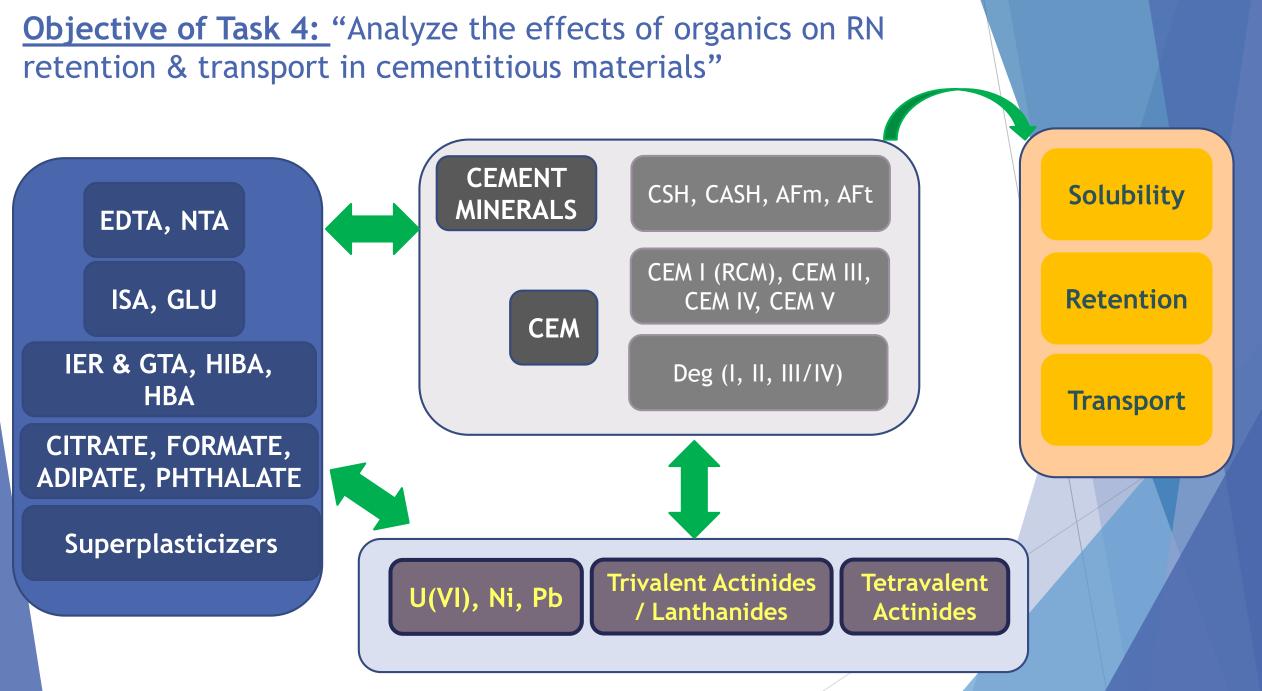


The project leading to this application has received funding from the European Union's Horizon 2020 research and innovation programme under grant agreement No 847593.

## Task 4 - Radionuclide-organic-cement-interactions

#### Main objectives:

- Improving the knowledge on organic-radionuclide complexes mobility in cement-based systems
- Studying the competition or synergetic effect in ternary systems (i.e. organic(s)/radionuclide/cement)
- Providing a mechanistic understanding of radionuclide interactions and quantitative transfer data in various cementitious environments



RCM: reference cement material

#### **Dedicated experiments using a CEM I - RCM**

Cement : CEM I 45.5R, CEMEX Prachovice, Czech republic;
HCP : water/cement = 0.45 - curing time > 28 days - portlandite saturated solution
→ 12 participants will include RCM in their work program: retention and/or transfer experiments



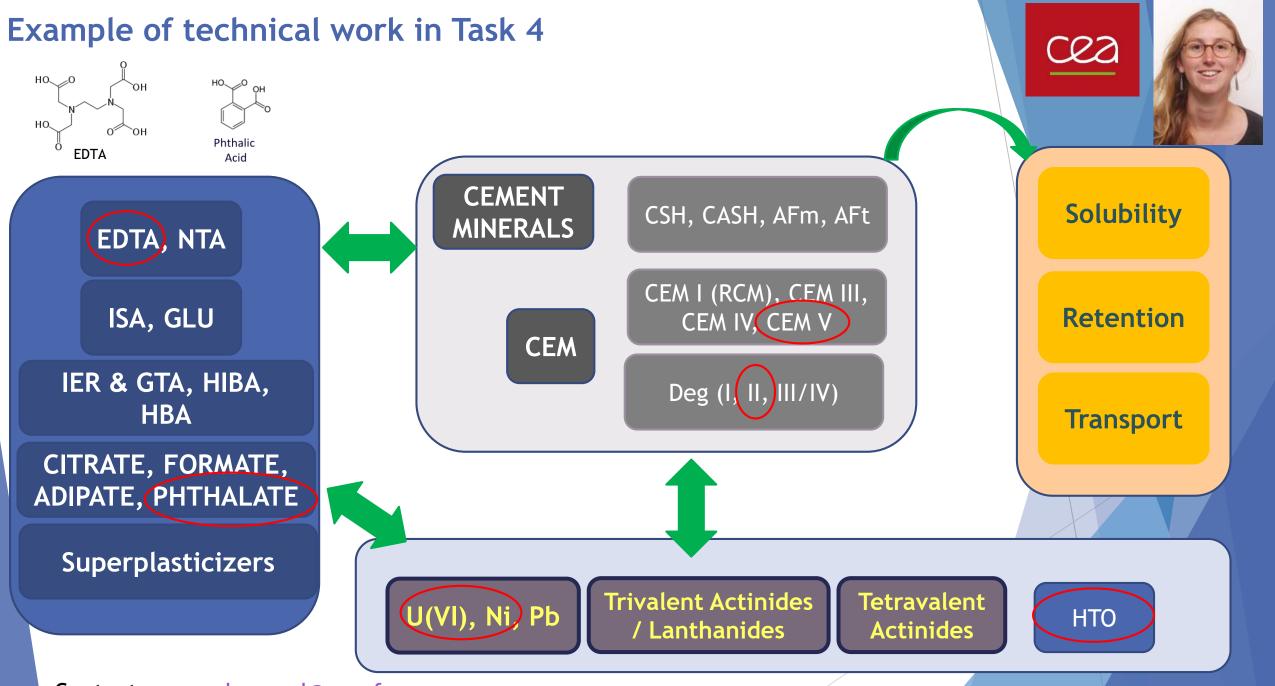
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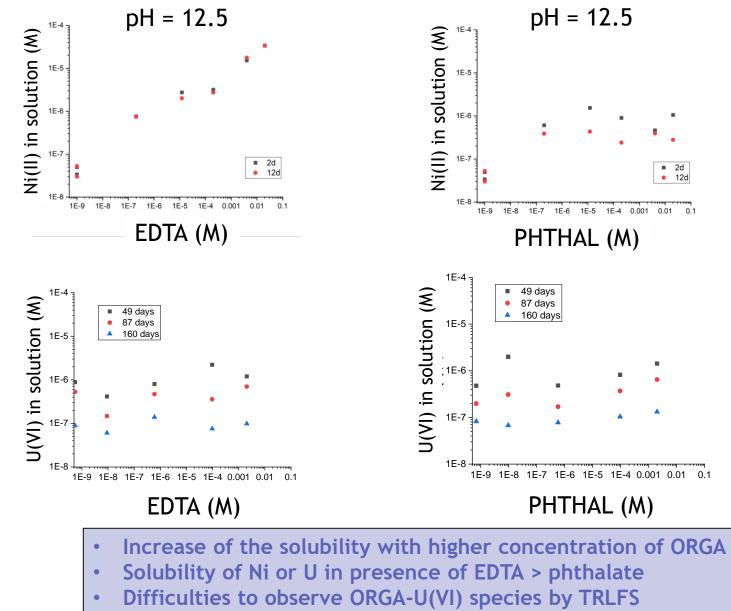
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RCM: reference cement material



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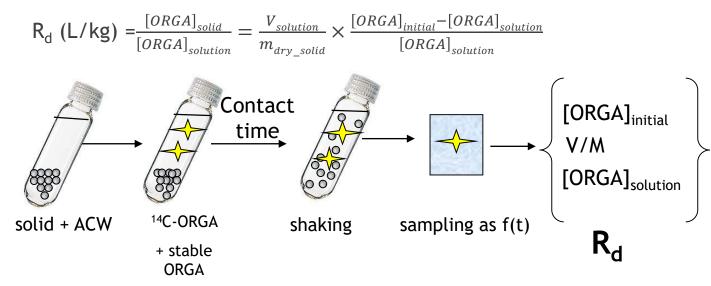
#### Step 1: operational solubility to size migration experiments



• [ORGA]<sub>ini</sub> for diffusion = 10<sup>-2</sup>M

# Step 2: Retardation factor estimation from batch sorption experiments

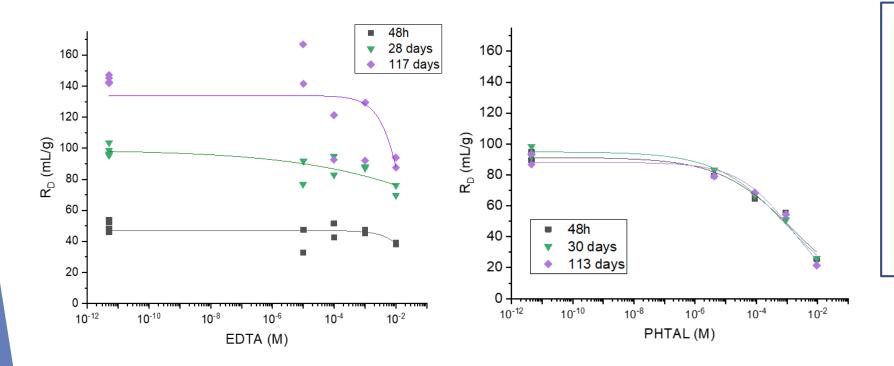


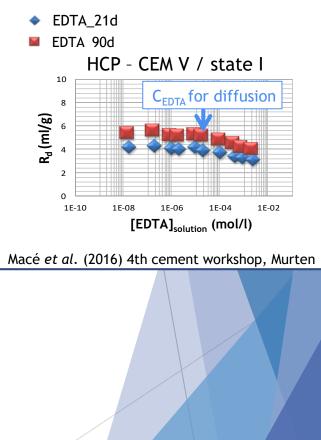


- Crushed HCP (CEM V/A, Water/Cement = 0.40) in suspension in ACW (pH 12.5)
- Use of <sup>14</sup>C-labelled species of EDTA and phthalate as ORGA radiotracers → retention in binary mode (Task3)
- Use of stable ORGA +  $^{63}$ Ni or U  $\rightarrow$  retention in ternary mode (Task 4)
- $_{\odot}$  [ORGA]<sub>initial</sub> = from 10<sup>-5</sup> to 10<sup>-2</sup> mol/L and [RN]<sub>initial</sub> < solubility
- Contact time = from 48h to >3 months
- $_{\odot}$   $\,$  LSC technique to measure  $^{14}\text{C}$  or  $^{63}\text{Ni}$  in solution, ICP-MS for U
- $_{\circ}$   $\,$  Usual analytical techniques to measure major species in solution

#### Determination of R<sub>d</sub> values as a function of [ORGA] and [RN]

#### Step 2: first results for binary systems with HCP state II

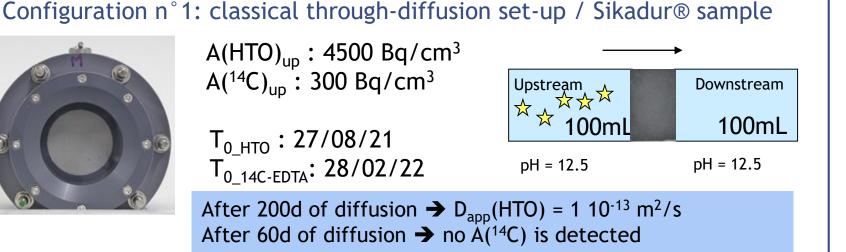




- $R_d(EDTA) > R_d(PHTHAL)$  for higher contact time
- Sorption kinetics :  $R_d(EDTA) \nearrow$  with time
- Strong decrease of  $R_d$  for [ORGA] > 10<sup>-5</sup>M;
- 1 site Langmuir isotherm;
- $[ORGA]_{ini}$  for diffusion =  $10^{-2}M$
- Ternary systems: on going experiments with Ni and U

## Step 3: On going diffusion experiments





Configuration n°2: optimized through-diffusion set-up / HCP sample - State II



A(HTO)<sub>up</sub>: 4500 Bq/cm<sup>3</sup> A(<sup>14</sup>C-EDTA)<sub>up</sub>: 5000 Bq/cm<sup>3</sup>

T<sub>0 HTO</sub>: 27/08/21

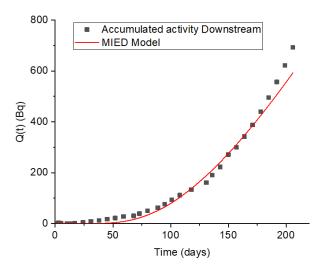
T<sub>0\_14C-EDTA</sub>: 15/12/21

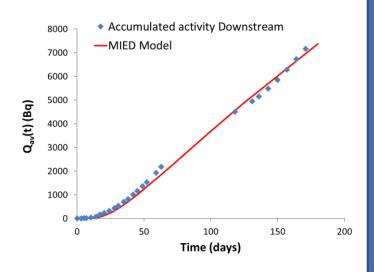


Upstream = dowstream  $\approx 10 \text{mL}$ 

pH = 12.5

After 200d of diffusion  $\rightarrow D_{app}(HTO) = 6.3 \ 10^{-13} \ m^2/s$ After 90d of diffusion  $\rightarrow$  no A(<sup>14</sup>C) is detected





## **Conclusive remarks**

ORGA: PHTALATE; EDTA <u>CEM:</u> HCP (CEM I, CEM V), deg (I, II), CASH, CSH RN: U(VI), Ni(II)

- Solubility
  - In portlandite saturated solution (pH = 12.5):
    - Increase of the solubility of Ni(II) with higher concentration of EDTA
    - Solubility of U(VI) or Ni(II) in presence of EDTA > phthalate
  - In CASH equilibrium solution (pH= 10.3-11.3) = solubility limit of U(VI) is c.a. 4.10<sup>-7</sup> M (data to be confirmed)

#### Retention

- For HCP CEM V state II:
  - $R_d$  of U(VI) ~ 6-30·10<sup>4</sup> L/kg + decrease of  $R_d$  with increasing [U(VI)];
  - $R_d$  (PHTAL/EDTA) < 1.5.10<sup>2</sup> L/kg + decrease of  $R_d$  with increasing [ORGA];
  - For [EDTA] up to  $10^{-2}$  mol·L<sup>-1</sup> decreases the resulting R<sub>d</sub> of Ni(II) : from  $10^3$  to < 1 L/kg;
- For HCP CEM I state II:
  - For [EDTA] =  $5 \cdot 10^{-5}$  mol·L<sup>-1</sup> no significant effect on the resulting R<sub>d</sub> of U(VI) ~  $10^4$  L/kg;
- For CSH (Ca/Si = 1),  $[U]_{ini} = 7 \cdot 10^{-8} \text{ mol} \cdot L^{-1} \rightarrow R_d$  values are higher in comparison with sorption on CEM I
- For CSH (Ca/Si = 1.27),  $[U]_{ini} = 5 \cdot 10^{-5} \text{ mol} \cdot L^{-1} \rightarrow \text{slight decrease of } R_d \text{ values with increasing [EDTA]};$

#### Diffusion

- Only HTO diffusion data in HCP state II are available so far
- Due to high  $R_d$  values  $\rightarrow$  only in-diffusion for RN-ORGA can occur, as expected

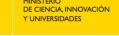
→ Puzzle is still not fully completed to be able to predict RN-ORGA interactions



## Overview of Task4 technical work contributions in NUWCEM











Karlsruher Institut für Technologie







Sorption of organics and uranium to CASH  $\rightarrow$  presented by James D. Begg

<sup>233</sup>U sorption on portlandite and the effect of organic compounds → presented by Oscar Almendros

**Degradation products** of UP2 filter aid material and their impact on radionuclide retention in cementitious systems

➔ presented by Xavi Gaona

Impact of formate, citrate and gluconate on the uptake of radionuclides by cement: study of the binary and ternary systems cement-L and cement-RN-L  $\rightarrow$  presented by Rosa E. Guidone

**Ni-63** interaction with cement material in the presence of organic compounds → presented by Petr Vercernik

Interaction of **europium** with **cementitious** materials in the presence of organic substance

→ presented by Marta Buresova

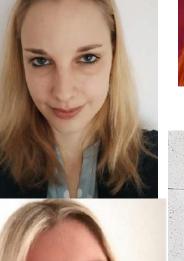
Sorption of uranium and lead on cement materials

→ presented by Jana Kittnerova

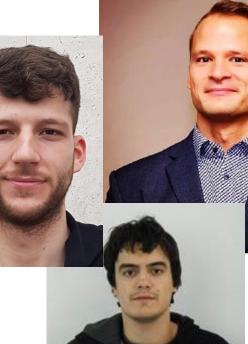
## Focus on young researchers involved in Task 4







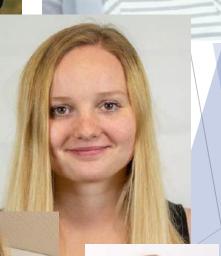




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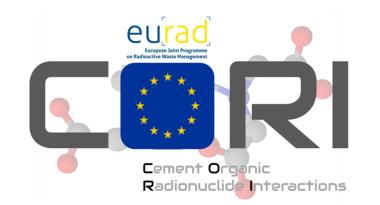
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# Thank you for your attention !!!



#### October23-25, 2019, Barcelona, Spain

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# Mechanisms and Modelling of Waste/Cement Interactions 6<sup>th</sup> International Workshop

Combined (back to back) with

the final CORI (eurad) meeting

Save the data: November 20 to 24<sup>th</sup> 2023 in Prague

Detailed information will follow soon

